# **Course Summary Solar Power for Africa**

Gave a little bit of information about Africa

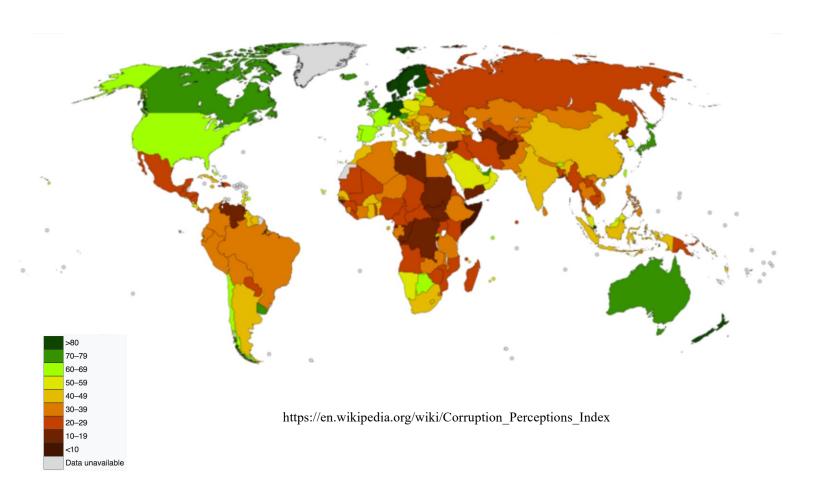
Went through some possible technologies that could be implemented to improve life

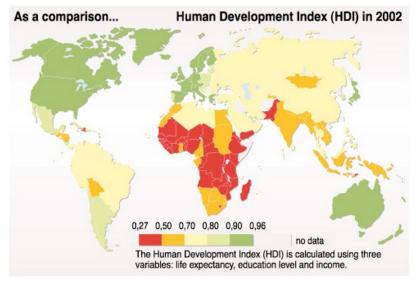
Looked into your possible roles in using what you have learned to try to impact the world

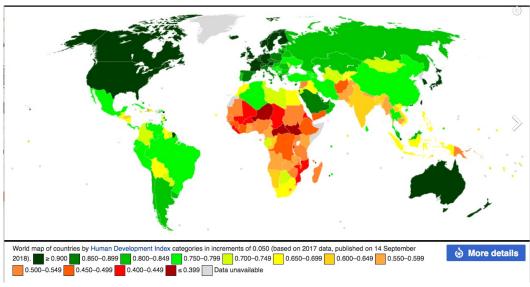
# Most of Africa is near the equator, this distorts the size in a 2d projection.

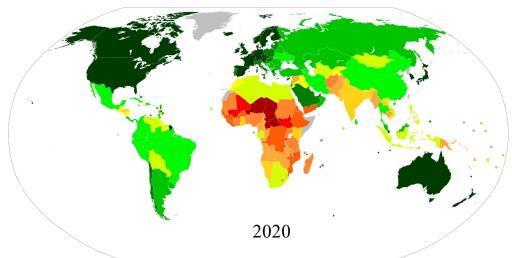


Corruption Index
"the misuse of public power for private benefit"

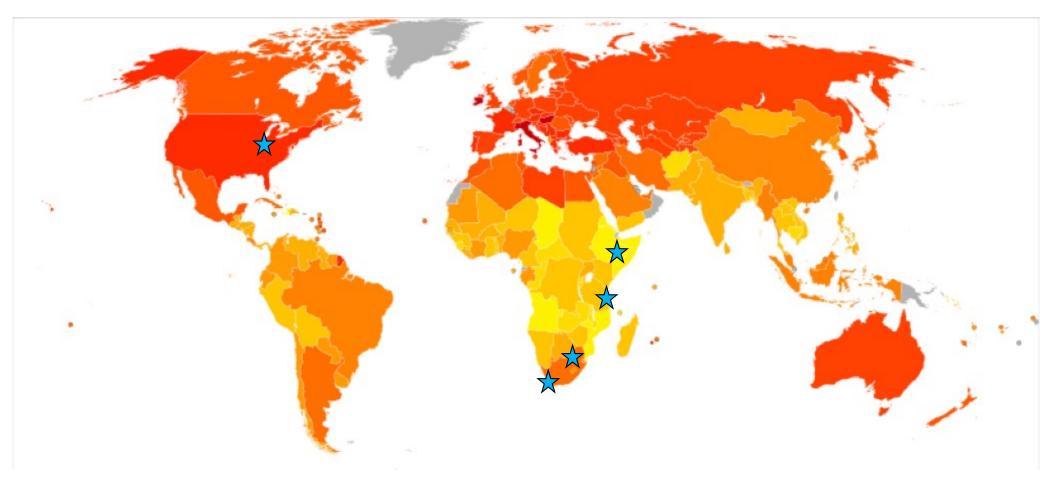






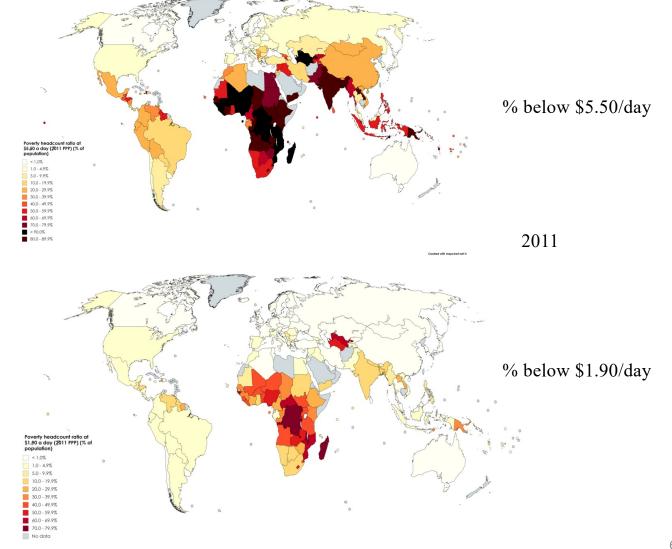


# **Energy Consumption in the World**



## **Poverty**

One cup of rice ~ \$0.22 One cup of beans ~ \$0.30 A wage of \$2/day is a survival wage Typically 12 hour work 7 days a week (Possibly 10 on Sunday)



6

# The world's largest cities in 2100

City

Population (2100)

From 2018 2 in Africa 13 drop

According to the report, human geography will look completely unfamil projected for 2100:

Country

	Population (2100	) City	Country	
#1	88.3 million	Lagos	Nigeria	None in US
#2	83.5 million	Kinshasa	DRC	None in China
#3	73.7 million	Dar Es Salaan	nTanzania	None in Europe
#4	67.2 million	Mumbai	India	Trone in Europe
#5	57.3 million	Delhi	India	10:
#6	56.6 million	Khartoum	Sudan	13 in Africa
#7	56.1 million	Niamey	Niger	
#8	54.3 million	Dhaka	Bangladesh	
#9	52.4 million	Kolkata	India	
#10	50.3 million	Kabul	Afghanistan	
#11	49.1 million	Karachi	Pakistan	
#12	46.7 million	Nairobi	Kenya	
#13	41.4 million	Lilongwe	Malawi	
#14	40.9 million	Blantyre City	Malawi	
#15	40.5 million	Cairo	Egypt	
#16	40.1 million	Kampala	Uganda	
	40.0 million	Manila	Philippines	
	37.7 million	Lusaka	Zambia	
	36.4 million	Mogadishu	Somalia	
#20	35.8 million	Addis Ababa	Ethiopia	

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a		•	•	•	•	•	•	•	•	•	•	•	•	•
	1	Tokyo	<ul><li>Japan</li></ul>	37,400,068	Metropolis prefecture	13,515,271	2,191	6,169 [14]	37,274,000	13,452	2,771 [15]	39,105,000	8,231	4,751 [e]
illi	2	Delhi	India	28,514,000	Capital City	16,753,235	1,484	11,289 [16]	29,000,000	3,483	8,326 [17]	31,870,000	2,233	14,272 [f]
	3	Shanghai _	China	25,582,000	Municipality	24,870,895	6,341	3,922 [18][19]	N/A	N/A	N/A	22,118,000	4,069	5,436 [g]
	4	São Paulo 🔴	◆ Brazil	21,650,000	Municipality	12,252,023	1,521	8,055 [20]	21,734,682	7,947	2,735 [21]	22,495,000	3,237	6,949 [h]
	5	Mexico City	■•■ Mexico	21,581,000	City-state	9,209,944	1,485	6,202 [22]	21,804,515	7,866	2,772 [23]	21,505,000	2,385	9,017
ι	6	Cairo	Egypt	20,076,000	Urban governorate	9,500,000	3,085	3,079 [24]	N/A	N/A	N/A	19,787,000	2,010	9,844
e	7	Mumbai	India	19,980,000	Municipality	12,478,447	603	20,694 [25]	24,400,000	4,355	5,603 [26]	22,186,000	1,008	22,010 [27][i]
	8	Beijing	China	19,618,000	Municipality	21,893,095	16,411	1,334 [18][19]	N/A	N/A	N/A	19,437,000	4,172	4,659
	9	Dhaka	Bangladesh	19,578,000	Capital city	8,906,039	338	26,349 [28][29]	14,543,124 [30]	N/A	N/A	16,839,000	456	36,928
	10	Osaka 🛑	<ul><li>Japan</li></ul>	19,281,000	Designated city	2,725,006	225	12,111 [14]	19,303,000	13,228	1,459 [15]	15,490,000	3,020	5,129 [j]
	11	New York	United States	18,819,000	City	8,804,190	778	11,316 [31]	20,140,470	12,093	1,665 [32]	23,582,649	34,493	684 [k]
	12	Karachi	© Pakistan	15,400,000	Metropolitan city	14,910,352	3,530	4,224 [33][34]	16,051,521	3,780	4,246 [35]	15,292,000	1,044	14,648 [36]
	13	Buenos Aires	Argentina	14,967,000	Autonomous city	3,054,300	203	15,046 [37]	12,806,866 [38]	N/A	N/A	16,216,000	3,222	5,033
	14	Chongqing	China	14,838,000	Municipality	32,054,159	82,403	389 [39][19]	N/A	N/A	N/A	8,261,000	1,536	5,378
	15	Istanbul	C- Turkey	14,751,000	Metropolitan municipality	15,519,267	5,196	2,987 [40]	N/A	N/A	N/A	15,311,000	1,375	11,135
	16	Kolkata	India	14,681,000	Municipality	4,496,694	205	21,935 [41]	14,035,959	1,851	7,583 [42]	18,698,000	1,352	13,830 [43]
	17	Manila	Philippines	13,482,000	Capital city	1,780,148	43	41,399 [44]	12,877,253	620	20,770 [44]	23,971,000	1,873	12,798 [I]

N/A

6,520,000

13.866,009 11.920

1,221

17,933

15,487,000

12,486,000 2,020

21,000,000 1,171

12,644,321

Area

UN 2018

population

13,463,000

13,293,000 Municipalit

13.215.000 Municipality

■ Nigeria

By the year 2100, it's estimated that 13 of the world's largest megacities will be located in Africa. Meanwhile, India will hold three of them – and there will be zero of them found in the Americas, China, or Europe.

18 Lagos

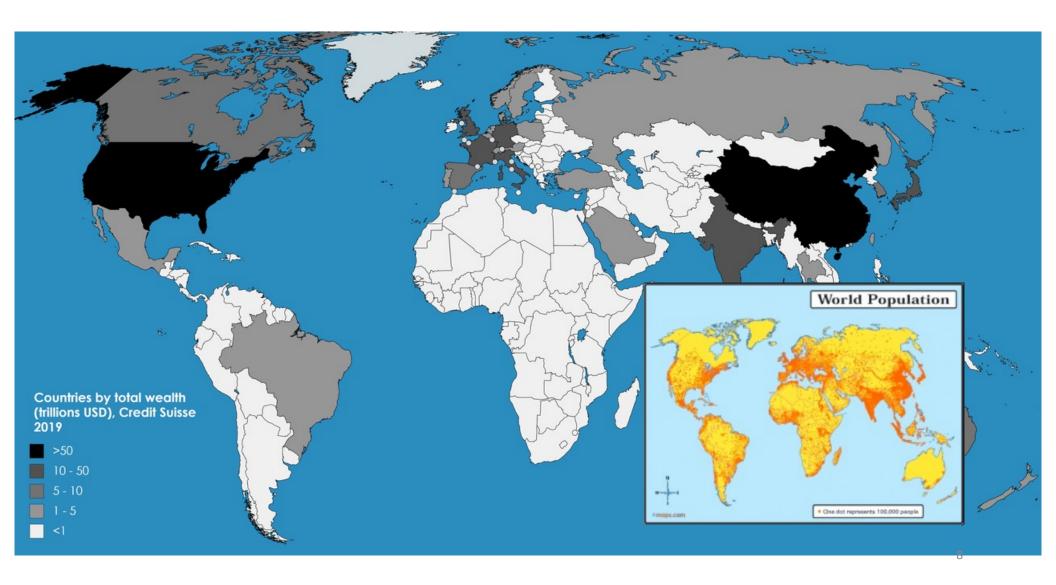




Figure 5.1 Remote and independent: a stand-alone system for a farmhouse.

- 1) PV Panels
- 2) Other sources of Power: Wind Turbine, Diesel or Gasoline Generator, Hydropower
- 3) Charge Controllers
- 4) Battery Bank
- 5) AC Inverter/Direct DC systems
- 6) Fuse box(es)
- 7) Appliances/Loads

50 Solar Street Lights Installed in Village in 2015 Improved Design Installed in 2016 by DDIT Students and Faculty (Commercialization was not followed through to date)

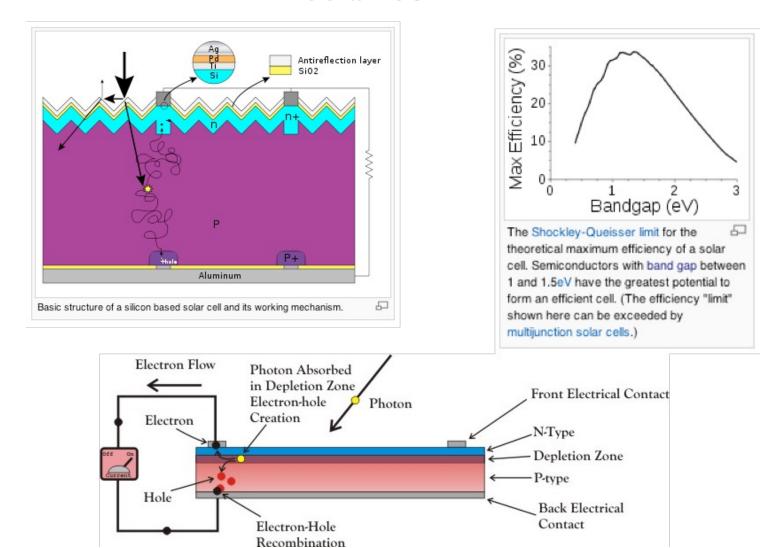


#### Power Ethiopia

-Photovoltaics are missing in Ethiopia high import tariffs no financing architectural adaption no technical support aversion to new technology no sales team

-Solve these issues using University Resources manufacture (assemble) PV panels train technologists/business people profit for faculty and students -HU/DDU area is unique for this opportunity

## Solar Cell



# **Oxidation/Reduction Metal Displacement Reaction**

#### Metal displacement [edit]

In this type of reaction, a metal atom in a compound (or in a solution) is replaced by an atom of another metal. For example, copper is deposited when zinc metal is placed in a copper(II) sulfate solution:

$$Zn(s)+CuSO_4(aq) \rightarrow ZnSO_4(aq) + Cu(s)$$

In the above reaction, zinc metal displaces the copper(II) ion from copper sulfate solution and thus liberates free copper metal.

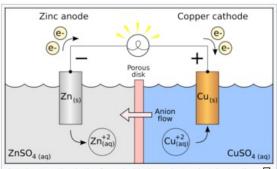
The ionic equation for this reaction is:

$$Zn + Cu^{2+} \rightarrow Zn^{2+} + Cu$$

As two half-reactions, it is seen that the zinc is oxidized:

$$Zn \rightarrow Zn^{2+} + 2e^{-}$$
 oxidation

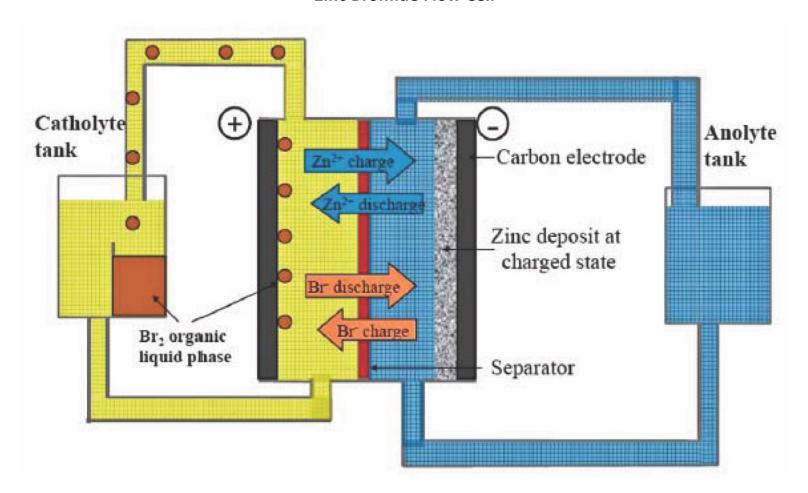
And the copper is reduced:



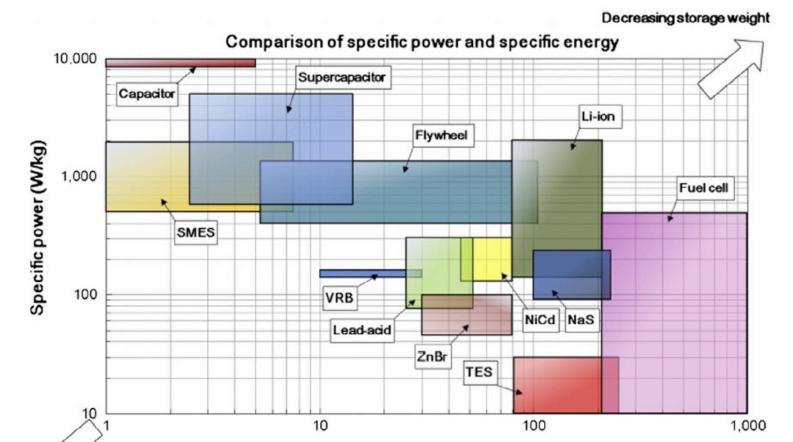
A redox reaction is the force behind an electrochemical cell
like the Galvanic cell pictured. The battery is made out of a zinc
electrode in a ZnSO<sub>4</sub> solution connected with a wire and a
porous disk to a copper electrode in a CuSO<sub>4</sub> solution.

#### 1.10 Volts for each cell

#### **Zinc Bromide Flow Cell**



# Comparing energy storage techniques



Specific energy (Wh/kg)

Increasing storage weight

# Lithium ion battery

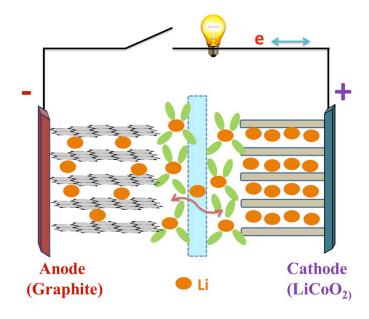
Intercalation reactions occur

Lithium-ions move into material lattices

Lithium ions shuttle back and forth

Graphite stores lithium when fully charged

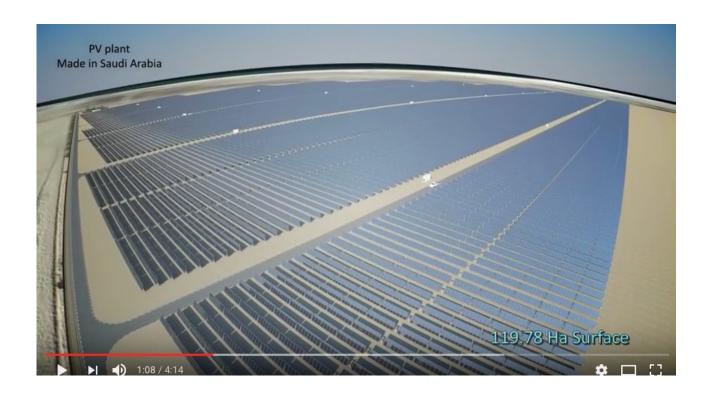
Cathode accepts lithium during discharge







#### King Abdullah's Desalinization Plant



#### <u>Seawatergreenhouse.com</u>

# A restorative approach to agriculture



Photo 3.1 The Gemasolar power tower near Sevilla (Spain)

Source: Torresol Energy.

## Key point

Molten-salts solar towers can generate electricity round the clock.

# Solar Chimney's

#### Enviromission (Australia)

http://www.enviromission.com.au/EVM/content/about\_companyprofile.

#### Solar Chimney Spain (Madrid)

http://www.youtube.com/watch?v=XCGVTYtJEFk





# Industrial Solar Water Heating Mek'elle Ethiopia

http://www.solarthermalworld.org/sites/gstec/files/Kahsay\_ISES% 20Kassel.pdf

No.	Factory	Process	Working Temperature (°C)	Consumption (m³/day)	Current Source of Energy
1.	Sheba	Skin Tanning	35	18.4	Furnace oil for
	Tannery	Skin Re-tanning	50	66.6	a steam boiler
	SALURISE:	Hide Tanning	40	29.3	
		Hide Re-tanning	65	27.0	
2.	Maichew	Glue	40	6.0	Furnace oil,
	Particleboard	preparation			fire wood
		Impregnation	55	1.2	
3.	Bahirdar	Pre-heater	60	36.0	Furnace oil for
	Textile	Washing	70	7.8	a steam boiler
	MIN CONTROL OF THE PARTY OF THE	Chemical	80	5.2	
		Preparation			
4.	Ashraf	Conditioning	85	6.0	Furnace oil for
	Edible Oil	Degumming	90	5.0	a steam boiler
		Neutralization	90	5.0	
		Washing	70	7.7	

#### Swamp Cooler (http://www.youtube.com/watch?v=6ooAAcsbf\_0)



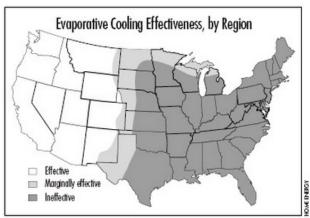


Figure 2. In areas of the United States where humidity tends to run high in the summertime, evaporative cooling is not the best way to stay cool; in the West, it's a very good choice.



#### A Simple Solar Ice Maker (Anthony Tong/Amanda)



Methanol/Carbon Based Absorption Refrigerator/Ice Maker



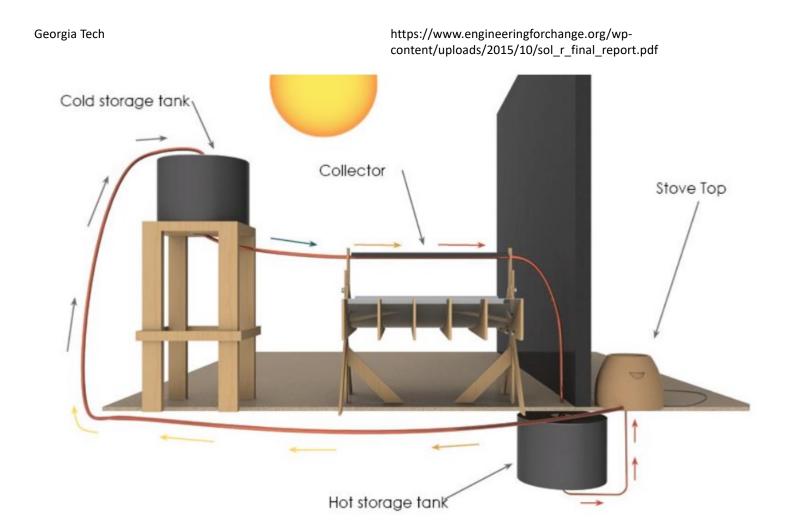


Figure VII.1. Depiction of the entire system

Go Sun Stove (http://www.kickstarter.com/projects/707808908/gosun-stove-portable-high-efficiency-solar-cooker)



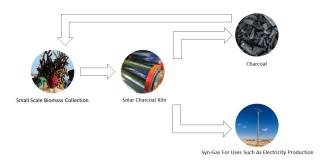
Go Sun Fusion

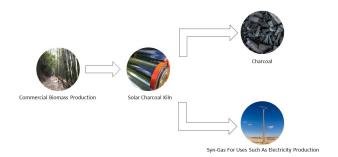




Figure 9 Picture of the prototype in its final version with the evacuated tube receiver.







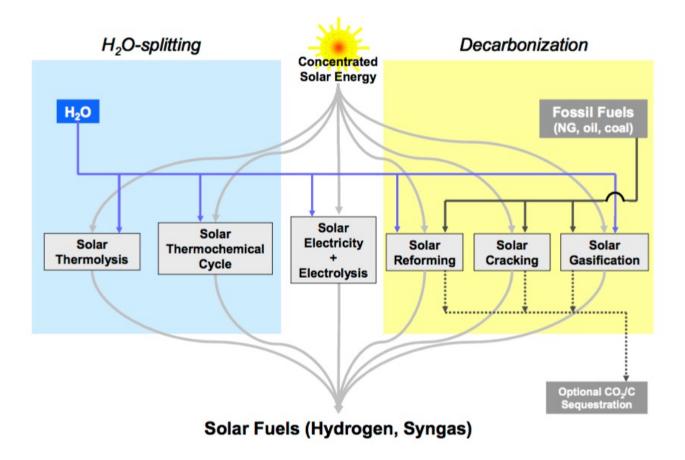
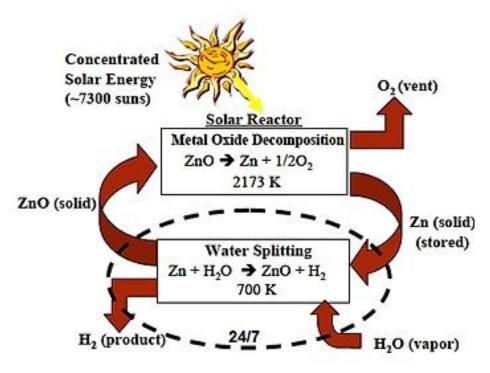


Fig. 2: Thermochemical routes for solar hydrogen production – Indicated is the chemical source of H<sub>2</sub>: H<sub>2</sub>O for the solar thermolysis and the solar thermochemical cycles; fossil fuels for the solar cracking, and a combination of fossil fuels and H<sub>2</sub>O for the solar reforming and gasification. For the solar decarbonization processes, optional CO<sub>2</sub>/C sequestration is considered. All of those routes involve energy consuming (endothermic) reactions that make use of concentrated solar radiation as the energy source of high-temperature process heat. Adapted from [1,2].

Zn => ZnO Cycle (http://en.wikipedia.org/wiki/Zinc%E2%80%93zinc\_oxide\_cycle)



UNLV (http://www.hydrogen.energy.gov/pdfs/review06/pd\_10\_weimer.pdf)

 $\underline{\textbf{PSI/ETHZ}} \ \textbf{(} \text{https://www.psi.ch/media/producing-pure-recycling-zinc-with-concentrated-solar-energy)} \\$ 



http://sfera.sollab.eu/downloads/Conferences/Sola Outline rPACES\_2012\_SFERA\_Meier.pdf

#### 100 kW Pilot Plant at MWSF

- Installation
- Commissioning

#### **Scientific Background**

- Solar ZnO dissociation at 2000 K
- Solar reactor technology

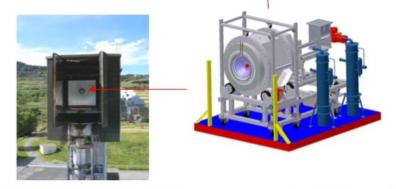
### **Experimental Results**

- Solar reactor experiments
- Flux measurements

## **Outlook / Acknowledgements**



CNRS 1 MW Solar Furnace Odeillo, France



# **Biomass/Syngas**

Steam Reforming Reaction: CH<sub>4</sub> to H<sub>2</sub> and CO

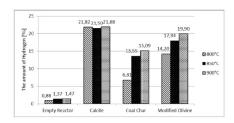
$$CH_4 + H_2O \rightarrow CO + 3H_2$$

At high temperatures (700 – 1100 °C) and in the presence of a metal-based catalyst (nickel), **steam** reacts with methane to yield carbon monoxide and hydrogen. ...

Water Shift Gas Reaction: CO to H<sub>2</sub>

$$CO + H_2O \rightarrow CO_2 + H_2$$

The shift reaction will operate with a variety of catalysts between 400°F and 90



 $Fig.~3.~Average~percentage~representation~of~hydrogen~for~the~studied~catalysts~at~t=800,~850~and~900^{\circ}C$ 

Fischer-Tropsch Reaction: H<sub>2</sub> and CO to Liquid Fuel

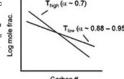
#### Reaction mechanism [edit]

The Fischer–Tropsch process involves a series of chemical reactions that produce a variety of hydrocarbons, ideally having the formula (C<sub>n</sub>H<sub>2n+2</sub>). The more useful reactions produce alkanes as follows:

$$(2n + 1) H_2 + n CO \rightarrow C_n H_{2n+2} + n H_2 O$$

where *n* is typically 10–20. The formation of methane (*n* = 1) is unwanted. Most of the alkanes produced tend to be straight-chain, suitable as diesel fuel. In addition to alkane formation, competing reactions give small amounts of alkenes, as well as alcohols and other oxygenated hydrocarbons.<sup>[4]</sup>

- Nickel (Ni) tends to promote methane formation, as in a methanation process; thus generally it is no desirable
- Iron (Fe) is relatively low cost and has a higher water-gas-shift activity, and is therefore more suitab a lower hydrogen/carbon monoxide ratio (H<sub>2</sub>/CO) syngas such as those derived from coal gasification
- Cobalt (Co) is more active, and generally preferred over ruthenium (Ru) because of the prohibitively cost of Ru
- In comparison to iron, Co has much less water-gas-shift activity, and is much more costly.



LowT	Sasol Arge	High T S	asol Synthol
• low C₁ = C, • low C₂ = C • low C₁₂ C₁ • 50-70% wt • 220-270°C • cc 0.87+	17.9 13.9 1x 51.7	- higher C <sub>1</sub> - C <sub>4</sub> - higher C <sub>5</sub> - C <sub>11</sub> - less C <sub>12</sub> -C <sub>19</sub> - low wax - 325 - 350°C - α: -0.7	43.0 40.0 7.0 4.0
<ul> <li>gasoline/di</li> <li>80° Cetane</li> <li>0-20 Octar</li> </ul>		<ul> <li>gasoline/diesel:</li> <li>50-60 Cetane #</li> <li>0-60 Octane #</li> </ul>	2:1

India Biogas Reactor (https://www.youtube.com/watch?v=9kKRdlAFuZw)



# Titania as a photocatalyst

- Irradiation of semiconductors having a band gap (2 4 eV) with UV light energy  $\geq$  energy of the band gap  $E_g$
- Generation of charge carriers

$$TiO_2 \xrightarrow{hv} TiO_2 (e^- + h^+)$$

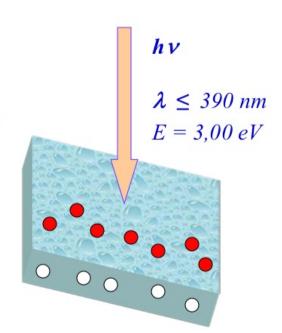
• Formation of active radicals ( $OH^*$ ,  $O_2^*$ )

$$OH^- + h^+ \longrightarrow OH^-$$

$$O^{2-} + e^{-} \longrightarrow {}^{\bullet}O^{2-}$$

• Recombination process

$$e^- + h^+ \longrightarrow Heat$$



#### **Solar Thermolysis**

#### Titania as a photocatalyst

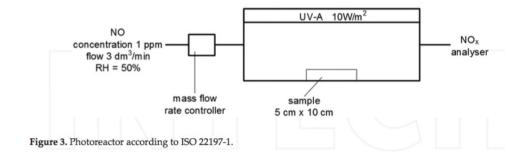


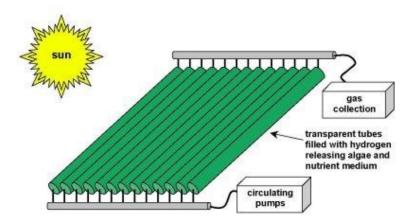


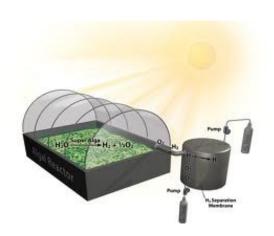
Figure 4. Separate parking lanes at the Leien of Antwerp with photocatalytic pavement blocks [15].

Spirulina Algae in West Africa (https://www.youtube.com/watch?v=CxSA5iiGgiY)



#### Simple schematic for biological hydrogen production



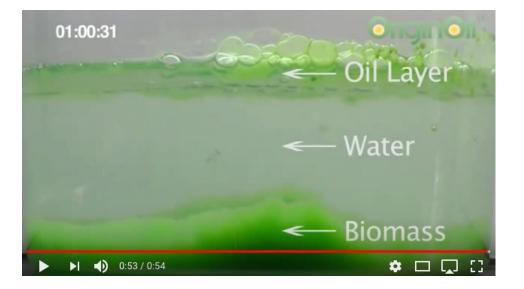




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#### Algae oil extraction

https://www.youtube.com/watch?v=aEZTAJNIxjQ



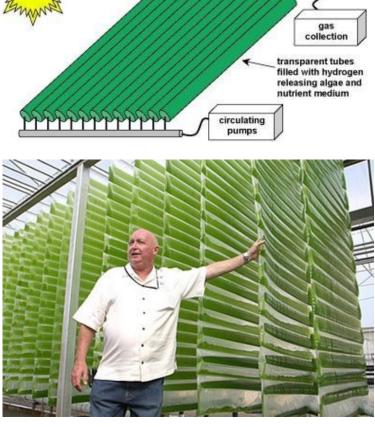
#### Algae Tower

#### Simple schematic for biological hydrogen production



<u>Hydrogen from Algae</u> (http://www.youtube.com/watch?v= Or\_F6qC0sK4)

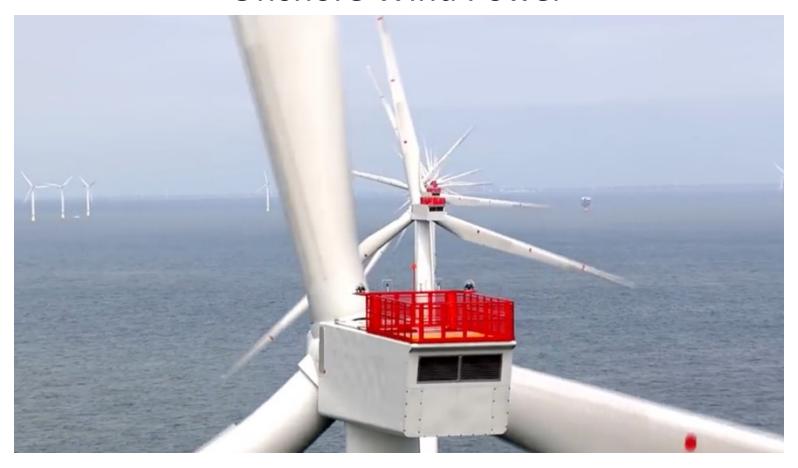
<u>Hydrogen from Algae Imperial College</u> (https://www.youtube.com/watch?v=OFByDMRbucs)



#### <u>Vertical Farming in Cincinnati</u>



# Offshore Wind Power



# **Course Summary Solar Power for Africa**

Hopefully, you learned something about the situation faced by most people on earth, more than half live on less than \$10 per day, 1 billion live on less than \$1/day about 10%. Africa has the poorest countries so it could be a place to start.

We are technologists. Our skill set is technological solutions to economic development and improvement of quality of life. Hopefully you got some ideas on how this could be reasonably implemented.

China thinks that the future is in Africa as evidenced by their investment \$110 billion in 2019 or 20% of Africa's economic growth. We should consider the business opportunities especially those that could benefit both parties in a partnership.